

12

EUROPEAN PATENT APPLICATION

21 Application number: 87311333.6

51 Int. Cl.⁴: **C 08 F 4/02**

C 08 F 4/64, C 08 F 10/00

22 Date of filing: **22.12.87**

30 Priority: 26.12.86 JP 308536/86

43 Date of publication of application:
06.07.88 Bulletin 88/27

64 Designated Contracting States: **BE DE FR GB NL**

71 Applicant: **TOA NENRYO KOGYO KABUSHIKI KAISHA**
1-1 Hitotsubashi, 1-Chome Chiyoda-Ku
Tokyo 100 (JP)

72 Inventor: **Murata, Masahide**
4-2-233 Nishitsurugaoka 1-chome
Ooi-machi Iruma-gun Saitama-ken (JP)

Imai, Masatumi
4-3-342 Nishitsurugaoka 1-chome
Ooi-machi Iruma-gun Saitama-ken (JP)

Furuhashi, Hiroyuki
4-6 Nishitsurugaoka 1-chome
Ooi-machi Iruma-gun Saitama-ken (JP)

Maruyama, Kouji
13-8 Chuou 2-chome
Kamifukuoka-shi Saitama-ken (JP)

Ueno, Hiroshi
398-1, Oosza Haneo Namegawa-machi
Hiki-gun Saitama-ken (JP)

74 Representative: **Dew, Melvyn John et al**
Exxon Chemical Limited Exxon Chemical Technology
Centre P.O. Box 1
Abingdon Oxfordshire, OX13 6BB (GB)

54 **Method of production of catalyst carrier for polymerization of olefin.**

57 A method for the production of a catalyst carrier for use in the polymerization of an olefin, which method comprises causing a magnesium-containing solid obtained by contact of (A) magnesium metal, (B) a halogenated hydrocarbon represented by the general formula, RX wherein R stands for an alkyl, aryl, or cycloalkyl group having 1 to 20 carbon atoms and X for a halogen atom, and (C) a compound of the general formula, $X^1_n M(OR^1)_{m-n}$ wherein X^1 stands for a hydrogen atom, a halogen atom, or a hydrocarbon group having 1 to 20 carbon atoms, M for a boron, carbon, silicon, aluminum, or phosphorus atom, R^1 for a hydrocarbon group having 1 to 20 carbon atoms, and m for the valency of said atom M , providing that $m > n \geq 0$ is satisfied, to contact (D) a halogen-containing alcohol.

EP 0 273 695 A1

Description**METHOD OF PRODUCTION OF CATALYST CARRIER FOR POLYMERIZATION OF OLEFIN****Field of the Invention**

- 5 This invention relates to a method for the production of a catalyst carrier for use in the polymerization of an olefin.

Description of the Prior Art

- 10 A magnesium-carried catalyst having a transition metal component such as titanium deposited on a magnesium compound-containing carrier and intended for use in the polymerization of an olefin has been known to the art. More often than not in the preparation of the catalyst, the carrier therefore is used after it has been mechanically pulverized. In this case, the produced catalyst for polymerization and consequently the polymer obtained therewith consist of particles which are devoid of uniformity in shape.

- 15 Recently, a few attempts are being made to improve the shape of particles of a carrier. For example, a method which comprises converting an aqueous magnesium chloride solution or dissolved magnesium chloride hydrate ($\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$) through spray drying into spherical particles (Japanese Patent Application Disclosure SHO 49(1974)-65,999, SHO 52(1977)-38,590, and SHO 54(1979)-41,985), a method which comprises mutually contacting magnesium metal, a halogenated hydrocarbon, and an electron donor type compound such as an alcohol (Japanese Patent Application Disclosure SHO 51(1976)-64,586), a method which comprises causing an organic magnesium compound to react with an ester of orthosilicic acid, and a method which comprises causing mutual reaction of an ester of orthosilicic acid and a halogenated hydrocarbon (Japanese Patent Application Disclosure SHO 53(1978)-146,292) have been proposed. Indeed these methods are capable of improving the shapes of particles of a carrier and a catalyst to some extent. The catalysts they produce, however, are not satisfactory in terms of activity.

- 25 The present inventors formerly found that a magnesium-containing solid consisting of particles uniform in the shape and effectively serving as a carrier for an olefin polymerization catalyst is obtained by causing contact of magnesium metal, a halogenated hydrocarbon, and a compound of the general formula, $\text{X}_m\text{C}(\text{OR})_{4-m}$ (Japanese Patent Application Disclosure SHO 56(1981)-125,407). The catalyst prepared by using this solid, however, has an unsolved problem concerning the activity for polymerization.

30

Problem for Solution by the Invention

An object of this invention is to provide a catalyst carrier for the polymerization of an olefin, excelling in the quality of particles and exhibiting efficiency in polymerization on a practical level.

Invention for Solution of the Problem

- 35 After a diligent study, the present inventors have found that the object of this invention can be accomplished by a magnesium-containing solid produced by contact of the magnesium-containing solid disclosed in Japanese Patent Application Disclosure SHO 56(1981)-125,407 and Japanese Patent Application Disclosure SHO 53(1978)-146,292 mentioned above with a halogen-containing alcohol. The present invention has been

40

Summary of the Invention

- 45 The essence of the present invention resides in a method for the production of a catalyst carrier, and the catalyst carrier obtained therefrom, for use in the polymerization of an olefin, which method comprises causing a magnesium-containing solid obtained by contact of (A) magnesium metal, (B) a halogenated hydrocarbon represented by the general formula, RX , wherein R stands for an alkyl, aryl, or cycloalkyl group having 1 to 20 carbon atoms and X for a halogen atom, and (C) a compound of the general formula, $\text{X}_n^1\text{M}(\text{OR}^1)_{m-n}$, wherein X^1 stands for a hydrogen atom, a halogen atom, or a hydrocarbon group having 1 to 20 carbon atoms, M for a boron, carbon, silicon, aluminium, or phosphorus atom, R^1 for a hydrocarbon group having 1 to 20 carbon atoms, and m for the valency of said atom M, providing that $m > n \geq 0$ to contact (D) a halogen-containing alcohol.

50

Raw Materials for Preparation of Carrier**(A) Magnesium Metal**

- 55 Though this invention does not discriminate magnesium metal on account of its form, magnesium metal is used particularly advantageously herein in the form of powder or chips. Preparatory to actual use herein, the magnesium metal is desired to be washed with an inactive hydrocarbon such as, for example, an aliphatic, alicyclic, or aromatic hydrocarbon having 6 to 8 carbon atoms and then dried by heating in the presence of an

60

inactive gas such as nitrogen.

(B) Halogenated Hydrocarbon

Of the halogenated hydrocarbons represented by the general formula, RX, particularly desirable are chlorinated or brominated hydrocarbons containing as the substituent, R; an alkyl, aryl, or cycloalkyl group having 1 to 8 carbon atoms. To be concrete, such particularly desirable halogenated hydrocarbons include methyl, ethyl, isopropyl, n-butyl, n-octyl, and cyclohexyl chlorides bromides, chlorobenzene, and o-chlorotoluene, for example.

(C) Compound of General Formula, $X \frac{1}{n} M(OR^1)_{m-n}$

In the formula, the symbols M, X^1 , R^1 , m, and n have the same meanings as defined above. The substituent, X^1 , may be a halogen-substituted hydrocarbon having 1 to 20 carbon atoms. Where X^1 is a hydrocarbon group, X^1 and R^1 may be equal or unequal. Hereinafter, the compounds of the aforementioned general formula will be referred to simply as "alkoxy compounds."

As examples of the hydrocarbon group, there can be cited alkyl groups such as methyl, ethyl, propyl, i-propyl, butyl, amyl, hexyl, octyl, 2-ethylhexyl, and decyl, cycloalkyl groups such as cyclopentyl, cyclohexyl, and methylcyclohexyl, alkenyl groups such as allyl, propenyl, and butynyl, aryl groups such as phenyl, tolyl, and xylyl, and aralkyl groups such as phenethyl and 3-phenylpropyl. Among other hydrocarbon groups enumerated above, alkyl groups of 1 to 10 carbon atoms prove to be particularly advantageous. Illustrative examples of the alkoxy compounds follow:

1. Compounds Using a Carbon Atom as the Substituent, M

The compounds of the formula, $C(OR^1)_4$, including $C(OCH_3)_4$, $C(OC_2H_5)_4$, $C(OC_3H_7)_4$, $C(OC_4H_9)_4$, $C(O-i-C_4H_9)_4$, $C(OC_6H_{13})_4$, and $C(OC_8H_{17})_4$; the compounds of the formula, $X^1C(OR^1)_3$, including $HC(OCH_3)_3$, $HC(OC_2H_5)_3$, $HC(OC_3H_7)_3$, $HC(OC_4H_9)_3$, $HC(O-i-C_4H_9)_3$, $HC(OC_6H_{13})_3$, $HC(OC_8H_{17})_3$, $HC(OC_6H_5)_3$, $CH_3C(OCH_3)_3$, $CH_3C(OC_2H_5)_3$, $C_2H_5C(OCH_3)_3$, $C_2H_5C(OC_2H_5)_3$, $C_6H_{11}C(OC_2H_5)_3$, $C_6H_5C(OCH_3)_3$, $C_6H_5C(OC_2H_5)_3$, $C_6H_5C(OC_3H_7)_3$, $C_7H_7C(OC_2H_5)_3$, $C_8H_9C(OC_2H_5)_3$, $CH_2BrC(OC_2H_5)_3$, $CH_2ClC(OC_2H_5)_3$, $CH_3CHBrC(OC_2H_5)_3$, $CH_3CHClC(OC_2H_5)_3$, $ClC(OCH_3)_3$, $ClC(OC_2H_5)_3$, $ClC(OC_3H_7)_3$, $ClC(O-i-C_4H_9)_3$, $ClC(OC_6H_{17})_3$, $ClC(OC_6H_5)_3$, and $BrC(OC_2H_5)_3$; and the compounds of the formula, $X \frac{1}{2} C(OR^1)_2$, including $CH_3CH(OCH_3)_2$, $CH_3CH(OC_2H_5)_2$, $CH_2(OCH_3)_2$, $CH_2(OC_2H_5)_2$, $CH_2ClCH(OC_2H_5)_2$, $CH_2Cl_2CH(OC_2H_5)_2$, $CCl_3CH(OC_2H_5)_2$, $CH_2BrCH(OC_2H_5)_2$, $CH_2ICH(OC_2H_5)_2$, and $C_6H_5CH(OC_2H_5)_2$.

2. Compounds Using a Silicon Atom as the Substituent, M

The compounds of the formula, $Si(OR^1)_4$, including $Si(OCH_3)_4$, $Si(OC_2H_5)_4$, $Si(OC_4H_9)_4$, $Si(Oi-C_4H_9)_4$, $Si(OC_6H_{13})_4$, $Si(OC_8H_{17})_4$, $Si[O-CH_2CH(C_2H_5)C_4H_9]_4$, and $Si(OC_6H_5)_4$; the compounds of the formula, $RSi(OR^1)_3$, including $HSi(OC_2H_5)_3$, $HSi(OC_4H_9)_3$, $HSi(OC_6H_{13})_3$, $HSi(OC_6H_5)_3$, $CH_3Si(OCH_3)_3$, $CH_3Si(OC_2H_5)_3$, $CH_3Si(OC_4H_9)_3$, $C_2H_5Si(OC_2H_5)_3$, $C_4H_9Si(OC_2H_5)_3$, $C_6H_5Si(OC_2H_5)_3$, $C_2H_5Si(OC_6H_5)_3$, $ClSi(OCH_3)_3$, $ClSi(OC_2H_5)_2$, $ClSi(OC_3H_7)_3$, $ClSi(OC_6H_5)_3$, and $BrSi(OC_2H_5)_3$; the compounds of the formula, $R_2Si(OR^1)_2$, including $(CH_3)_2Si(OCH_3)_2$, $(CH_3)_2Si(OC_2H_5)_2$, $(CH_3)_2Si(OC_3H_7)_2$, $(C_2H_5)_2Si(OC_2H_5)_2$, $(C_6H_5)_2Si(OC_2H_5)_2$, $CH_3ClSi(OC_2H_5)_2$, $CHCl_2SiH(OC_2H_5)_2$, $CCl_3SiH(OC_2H_5)_2$, $CH_3BrSiH(OC_2H_5)_2$, and $CH_3SiH(OC_2H_5)_2$; and the compounds of the formula, R_3SiOR^1 , including $(CH_3)_3SiOCH_3$, $(CH_3)_3SiOC_2H_5$, $(CH_3)_3SiOC_4H_9$, $(CH_3)_3SiOC_6H_5$, $(C_2H_5)_3SiOC_2H_5$, and $(C_6H_5)_3SiOC_2H_5$.

3. Compounds Using a Boron Atom as the Substituent, M

The compounds of the formula, $B(OR^1)_3$, including $B(OC_2H_5)_3$, $B(OC_4H_9)_3$, $B(OC_6H_{13})_3$, and $B(OC_6H_5)_3$.

4. Compounds Using an Aluminum Atom as the Substituent, M

The compounds of the formula, $Al(OR^1)_3$, including $Al(OCH_3)_3$, $Al(OC_2H_5)_3$, $Al(OC_3H_7)_3$, $Al(Oi-C_3H_7)_3$, $Al(OC_4H_9)_3$, $Al(Ot-C_4H_9)_3$, $Al(OC_6H_{13})_3$, and $Al(OC_6H_5)_3$.

5. Compounds Using a Phosphorus Atom as the Substituent, M

The compounds of the formula, $P(OR^1)_3$, including $P(OCH_3)_3$, $P(OC_2H_5)_3$, $P(OC_4H_9)_3$, $P(OC_6H_{13})_3$, and $P(OC_6H_5)_3$.

(D) Halogen-Containing Alcohol

The term "halogen-containing alcohol" as used in the present invention means a compound obtained from a monohydric or polyhydric alcohol containing one or more hydroxyl groups in the molecular unit thereof by substituting a halogen atom for at least one freely selected hydrogen atom other than the hydrogen atom of the hydroxyl group. As examples of the halogen atom, there can be cited chlorine, bromine, iodine, and fluorine atoms. Among the halogen atoms cited above, the chlorine atom proves to be particularly advantageous.

As examples of the compounds of the foregoing definition, there can be cited 2-chloroethanol, 1-chloro-2-propanol, 3-chloro-1-propanol, 1-chloro-2-methyl-2-propanol, 4-chloro-1-butanol, 5-chloro-1-pentanol, 6-chloro-1-hexanol, 3-chloro-1,2-propane diol, 2-chlorocyclohexanol, 4-chlorobenzhydrol, (m,o,p)-chlorobenzyl alcohol, 4-chlorocatechol, 4-chloro-(m,o)-cresol, 6-chloro-(m,o)-cresol, 4-chloro-3,5-dimethyl phenol, chlorohydroquinone, 2-benzyl-4-chlorophenol, 4-chloro-1-naphthol, (m,o,p)-chlorophenol, p-chloro- α -methylbenzyl alcohol, 2-chloro-4-phenyl phenol, 6-chlorothymol, 4-chlororesorcinol, 2-bromoethanol,

3-bromo-1-propanol, 1-bromo-2-propanol, 1-bromo-2-butanol, 2-bromo-p-cresol, 1-bromo-2-naphthol, 6-bromo-2-naphthol, (m,o,p)-bromophenol, 4-bromoresorcinol, (m,o,p)-fluorophenol, p-iodophenol; 2,2-dichloroethanol, 2,3-dichloro-1-propanol, 1,3-dichloro-2-propanol, 3-chloro-1-(α -chloromethyl)-1-propanol, 2,3-dibromo-1-propanol, 1,3-dibromo-2-propanol, 2,4-dibromophenol, 2,4-dibromo-1-naphthol; 2,2,2-trichloroethanol, 1,1,1-trichloro-2-propanol, β,β,β -trichloro-tert-butanol, 2,3,4-trichlorophenol, 2,4,5-trichlorophenol, 2,4,6-trichlorophenol, 2,4,6-tribromophenol, 2,3,5-tribromo-2-hydroxy toluene, 2,3,5-tribromo-4-hydroxy toluene, 2,2,2-trifluoroethanol, α,α,α -trifluoro-m-cresol, 2,4,6-triiodophenol, 2,3,4,6-tetrachlorophenol, tetrachloro-hydroquinone, tetrachloro-bis-phenol A, tetrabromo-bis-phenol A, 2,2,3,3-tetrafluoro-1-propanol, 2,3,5,6-tetrafluorophenol, and tetrafluororesorcinol.

10

Procedure for Preparation of Carrier

The carrier mentioned above is obtained by causing a halogen-containing alcohol to come into contact with the product of contact of magnesium metal, a halogenated hydrocarbon, and an alkoxy compound.

(1) Contact Among Magnesium Metal, Halogenated Hydrocarbon, and Alkoxy Compound

The method for establishing contact among the three components is not specifically limited. Virtually any method can be adopted for this contact. This contact can be accomplished, for example, by (1) a method which comprises simultaneously placing the three components into mutual contact, (2) a method which comprises preparatorily allowing magnesium metal to come into contact with the halogenated hydrocarbon and subsequently allowing the product of this contact or a compound obtained in advance by the contact of the two components, such as, for example, any of the compounds represented by the formulas, ClMgCH_3 , ClMgC_2H_5 , ClMgC_3H_7 , ClMgC_4H_9 , $\text{ClMgC}_6\text{H}_{13}$, $\text{ClMgC}_8\text{H}_{17}$, BrMgC_2H_5 , BrMgC_4H_9 , $\text{BrMgC}_6\text{H}_{13}$, $\text{BrMgC}_8\text{H}_{17}$, $\text{BrMgC}_{10}\text{H}_{21}$, $\text{BrMgC}_{12}\text{H}_{25}$, $\text{BrMgC}_{14}\text{H}_{29}$, $\text{BrMgC}_{16}\text{H}_{33}$, $\text{BrMgC}_{18}\text{H}_{37}$, $\text{BrMgC}_{20}\text{H}_{41}$, $\text{BrMgC}_{22}\text{H}_{45}$, $\text{BrMgC}_{24}\text{H}_{49}$, $\text{BrMgC}_{26}\text{H}_{53}$, $\text{BrMgC}_{28}\text{H}_{57}$, $\text{BrMgC}_{30}\text{H}_{61}$, $\text{BrMgC}_{32}\text{H}_{65}$, $\text{BrMgC}_{34}\text{H}_{69}$, $\text{BrMgC}_{36}\text{H}_{73}$, $\text{BrMgC}_{38}\text{H}_{77}$, $\text{BrMgC}_{40}\text{H}_{81}$, $\text{BrMgC}_{42}\text{H}_{85}$, $\text{BrMgC}_{44}\text{H}_{89}$, $\text{BrMgC}_{46}\text{H}_{93}$, $\text{BrMgC}_{48}\text{H}_{97}$, $\text{BrMgC}_{50}\text{H}_{101}$, $\text{BrMgC}_{52}\text{H}_{105}$, $\text{BrMgC}_{54}\text{H}_{109}$, $\text{BrMgC}_{56}\text{H}_{113}$, $\text{BrMgC}_{58}\text{H}_{117}$, $\text{BrMgC}_{60}\text{H}_{121}$, $\text{BrMgC}_{62}\text{H}_{125}$, $\text{BrMgC}_{64}\text{H}_{129}$, $\text{BrMgC}_{66}\text{H}_{133}$, $\text{BrMgC}_{68}\text{H}_{137}$, $\text{BrMgC}_{70}\text{H}_{141}$, $\text{BrMgC}_{72}\text{H}_{145}$, $\text{BrMgC}_{74}\text{H}_{149}$, $\text{BrMgC}_{76}\text{H}_{153}$, $\text{BrMgC}_{78}\text{H}_{157}$, $\text{BrMgC}_{80}\text{H}_{161}$, $\text{BrMgC}_{82}\text{H}_{165}$, $\text{BrMgC}_{84}\text{H}_{169}$, $\text{BrMgC}_{86}\text{H}_{173}$, $\text{BrMgC}_{88}\text{H}_{177}$, $\text{BrMgC}_{90}\text{H}_{181}$, $\text{BrMgC}_{92}\text{H}_{185}$, $\text{BrMgC}_{94}\text{H}_{189}$, $\text{BrMgC}_{96}\text{H}_{193}$, $\text{BrMgC}_{98}\text{H}_{197}$, $\text{BrMgC}_{100}\text{H}_{201}$, $\text{BrMgC}_{102}\text{H}_{205}$, $\text{BrMgC}_{104}\text{H}_{209}$, $\text{BrMgC}_{106}\text{H}_{213}$, $\text{BrMgC}_{108}\text{H}_{217}$, $\text{BrMgC}_{110}\text{H}_{221}$, $\text{BrMgC}_{112}\text{H}_{225}$, $\text{BrMgC}_{114}\text{H}_{229}$, $\text{BrMgC}_{116}\text{H}_{233}$, $\text{BrMgC}_{118}\text{H}_{237}$, $\text{BrMgC}_{120}\text{H}_{241}$, $\text{BrMgC}_{122}\text{H}_{245}$, $\text{BrMgC}_{124}\text{H}_{249}$, $\text{BrMgC}_{126}\text{H}_{253}$, $\text{BrMgC}_{128}\text{H}_{257}$, $\text{BrMgC}_{130}\text{H}_{261}$, $\text{BrMgC}_{132}\text{H}_{265}$, $\text{BrMgC}_{134}\text{H}_{269}$, $\text{BrMgC}_{136}\text{H}_{273}$, $\text{BrMgC}_{138}\text{H}_{277}$, $\text{BrMgC}_{140}\text{H}_{281}$, $\text{BrMgC}_{142}\text{H}_{285}$, $\text{BrMgC}_{144}\text{H}_{289}$, $\text{BrMgC}_{146}\text{H}_{293}$, $\text{BrMgC}_{148}\text{H}_{297}$, $\text{BrMgC}_{150}\text{H}_{301}$, $\text{BrMgC}_{152}\text{H}_{305}$, $\text{BrMgC}_{154}\text{H}_{309}$, $\text{BrMgC}_{156}\text{H}_{313}$, $\text{BrMgC}_{158}\text{H}_{317}$, $\text{BrMgC}_{160}\text{H}_{321}$, $\text{BrMgC}_{162}\text{H}_{325}$, $\text{BrMgC}_{164}\text{H}_{329}$, $\text{BrMgC}_{166}\text{H}_{333}$, $\text{BrMgC}_{168}\text{H}_{337}$, $\text{BrMgC}_{170}\text{H}_{341}$, $\text{BrMgC}_{172}\text{H}_{345}$, $\text{BrMgC}_{174}\text{H}_{349}$, $\text{BrMgC}_{176}\text{H}_{353}$, $\text{BrMgC}_{178}\text{H}_{357}$, $\text{BrMgC}_{180}\text{H}_{361}$, $\text{BrMgC}_{182}\text{H}_{365}$, $\text{BrMgC}_{184}\text{H}_{369}$, $\text{BrMgC}_{186}\text{H}_{373}$, $\text{BrMgC}_{188}\text{H}_{377}$, $\text{BrMgC}_{190}\text{H}_{381}$, $\text{BrMgC}_{192}\text{H}_{385}$, $\text{BrMgC}_{194}\text{H}_{389}$, $\text{BrMgC}_{196}\text{H}_{393}$, $\text{BrMgC}_{198}\text{H}_{397}$, $\text{BrMgC}_{200}\text{H}_{401}$, $\text{BrMgC}_{202}\text{H}_{405}$, $\text{BrMgC}_{204}\text{H}_{409}$, $\text{BrMgC}_{206}\text{H}_{413}$, $\text{BrMgC}_{208}\text{H}_{417}$, $\text{BrMgC}_{210}\text{H}_{421}$, $\text{BrMgC}_{212}\text{H}_{425}$, $\text{BrMgC}_{214}\text{H}_{429}$, $\text{BrMgC}_{216}\text{H}_{433}$, $\text{BrMgC}_{218}\text{H}_{437}$, $\text{BrMgC}_{220}\text{H}_{441}$, $\text{BrMgC}_{222}\text{H}_{445}$, $\text{BrMgC}_{224}\text{H}_{449}$, $\text{BrMgC}_{226}\text{H}_{453}$, $\text{BrMgC}_{228}\text{H}_{457}$, $\text{BrMgC}_{230}\text{H}_{461}$, $\text{BrMgC}_{232}\text{H}_{465}$, $\text{BrMgC}_{234}\text{H}_{469}$, $\text{BrMgC}_{236}\text{H}_{473}$, $\text{BrMgC}_{238}\text{H}_{477}$, $\text{BrMgC}_{240}\text{H}_{481}$, $\text{BrMgC}_{242}\text{H}_{485}$, $\text{BrMgC}_{244}\text{H}_{489}$, $\text{BrMgC}_{246}\text{H}_{493}$, $\text{BrMgC}_{248}\text{H}_{497}$, $\text{BrMgC}_{250}\text{H}_{501}$, $\text{BrMgC}_{252}\text{H}_{505}$, $\text{BrMgC}_{254}\text{H}_{509}$, $\text{BrMgC}_{256}\text{H}_{513}$, $\text{BrMgC}_{258}\text{H}_{517}$, $\text{BrMgC}_{260}\text{H}_{521}$, $\text{BrMgC}_{262}\text{H}_{525}$, $\text{BrMgC}_{264}\text{H}_{529}$, $\text{BrMgC}_{266}\text{H}_{533}$, $\text{BrMgC}_{268}\text{H}_{537}$, $\text{BrMgC}_{270}\text{H}_{541}$, $\text{BrMgC}_{272}\text{H}_{545}$, $\text{BrMgC}_{274}\text{H}_{549}$, $\text{BrMgC}_{276}\text{H}_{553}$, $\text{BrMgC}_{278}\text{H}_{557}$, $\text{BrMgC}_{280}\text{H}_{561}$, $\text{BrMgC}_{282}\text{H}_{565}$, $\text{BrMgC}_{284}\text{H}_{569}$, $\text{BrMgC}_{286}\text{H}_{573}$, $\text{BrMgC}_{288}\text{H}_{577}$, $\text{BrMgC}_{290}\text{H}_{581}$, $\text{BrMgC}_{292}\text{H}_{585}$, $\text{BrMgC}_{294}\text{H}_{589}$, $\text{BrMgC}_{296}\text{H}_{593}$, $\text{BrMgC}_{298}\text{H}_{597}$, $\text{BrMgC}_{300}\text{H}_{601}$, $\text{BrMgC}_{302}\text{H}_{605}$, $\text{BrMgC}_{304}\text{H}_{609}$, $\text{BrMgC}_{306}\text{H}_{613}$, $\text{BrMgC}_{308}\text{H}_{617}$, $\text{BrMgC}_{310}\text{H}_{621}$, $\text{BrMgC}_{312}\text{H}_{625}$, $\text{BrMgC}_{314}\text{H}_{629}$, $\text{BrMgC}_{316}\text{H}_{633}$, $\text{BrMgC}_{318}\text{H}_{637}$, $\text{BrMgC}_{320}\text{H}_{641}$, $\text{BrMgC}_{322}\text{H}_{645}$, $\text{BrMgC}_{324}\text{H}_{649}$, $\text{BrMgC}_{326}\text{H}_{653}$, $\text{BrMgC}_{328}\text{H}_{657}$, $\text{BrMgC}_{330}\text{H}_{661}$, $\text{BrMgC}_{332}\text{H}_{665}$, $\text{BrMgC}_{334}\text{H}_{669}$, $\text{BrMgC}_{336}\text{H}_{673}$, $\text{BrMgC}_{338}\text{H}_{677}$, $\text{BrMgC}_{340}\text{H}_{681}$, $\text{BrMgC}_{342}\text{H}_{685}$, $\text{BrMgC}_{344}\text{H}_{689}$, $\text{BrMgC}_{346}\text{H}_{693}$, $\text{BrMgC}_{348}\text{H}_{697}$, $\text{BrMgC}_{350}\text{H}_{701}$, $\text{BrMgC}_{352}\text{H}_{705}$, $\text{BrMgC}_{354}\text{H}_{709}$, $\text{BrMgC}_{356}\text{H}_{713}$, $\text{BrMgC}_{358}\text{H}_{717}$, $\text{BrMgC}_{360}\text{H}_{721}$, $\text{BrMgC}_{362}\text{H}_{725}$, $\text{BrMgC}_{364}\text{H}_{729}$, $\text{BrMgC}_{366}\text{H}_{733}$, $\text{BrMgC}_{368}\text{H}_{737}$, $\text{BrMgC}_{370}\text{H}_{741}$, $\text{BrMgC}_{372}\text{H}_{745}$, $\text{BrMgC}_{374}\text{H}_{749}$, $\text{BrMgC}_{376}\text{H}_{753}$, $\text{BrMgC}_{378}\text{H}_{757}$, $\text{BrMgC}_{380}\text{H}_{761}$, $\text{BrMgC}_{382}\text{H}_{765}$, $\text{BrMgC}_{384}\text{H}_{769}$, $\text{BrMgC}_{386}\text{H}_{773}$, $\text{BrMgC}_{388}\text{H}_{777}$, $\text{BrMgC}_{390}\text{H}_{781}$, $\text{BrMgC}_{392}\text{H}_{785}$, $\text{BrMgC}_{394}\text{H}_{789}$, $\text{BrMgC}_{396}\text{H}_{793}$, $\text{BrMgC}_{398}\text{H}_{797}$, $\text{BrMgC}_{400}\text{H}_{801}$, $\text{BrMgC}_{402}\text{H}_{805}$, $\text{BrMgC}_{404}\text{H}_{809}$, $\text{BrMgC}_{406}\text{H}_{813}$, $\text{BrMgC}_{408}\text{H}_{817}$, $\text{BrMgC}_{410}\text{H}_{821}$, $\text{BrMgC}_{412}\text{H}_{825}$, $\text{BrMgC}_{414}\text{H}_{829}$, $\text{BrMgC}_{416}\text{H}_{833}$, $\text{BrMgC}_{418}\text{H}_{837}$, $\text{BrMgC}_{420}\text{H}_{841}$, $\text{BrMgC}_{422}\text{H}_{845}$, $\text{BrMgC}_{424}\text{H}_{849}$, $\text{BrMgC}_{426}\text{H}_{853}$, $\text{BrMgC}_{428}\text{H}_{857}$, $\text{BrMgC}_{430}\text{H}_{861}$, $\text{BrMgC}_{432}\text{H}_{865}$, $\text{BrMgC}_{434}\text{H}_{869}$, $\text{BrMgC}_{436}\text{H}_{873}$, $\text{BrMgC}_{438}\text{H}_{877}$, $\text{BrMgC}_{440}\text{H}_{881}$, $\text{BrMgC}_{442}\text{H}_{885}$, $\text{BrMgC}_{444}\text{H}_{889}$, $\text{BrMgC}_{446}\text{H}_{893}$, $\text{BrMgC}_{448}\text{H}_{897}$, $\text{BrMgC}_{450}\text{H}_{901}$, $\text{BrMgC}_{452}\text{H}_{905}$, $\text{BrMgC}_{454}\text{H}_{909}$, $\text{BrMgC}_{456}\text{H}_{913}$, $\text{BrMgC}_{458}\text{H}_{917}$, $\text{BrMgC}_{460}\text{H}_{921}$, $\text{BrMgC}_{462}\text{H}_{925}$, $\text{BrMgC}_{464}\text{H}_{929}$, $\text{BrMgC}_{466}\text{H}_{933}$, $\text{BrMgC}_{468}\text{H}_{937}$, $\text{BrMgC}_{470}\text{H}_{941}$, $\text{BrMgC}_{472}\text{H}_{945}$, $\text{BrMgC}_{474}\text{H}_{949}$, $\text{BrMgC}_{476}\text{H}_{953}$, $\text{BrMgC}_{478}\text{H}_{957}$, $\text{BrMgC}_{480}\text{H}_{961}$, $\text{BrMgC}_{482}\text{H}_{965}$, $\text{BrMgC}_{484}\text{H}_{969}$, $\text{BrMgC}_{486}\text{H}_{973}$, $\text{BrMgC}_{488}\text{H}_{977}$, $\text{BrMgC}_{490}\text{H}_{981}$, $\text{BrMgC}_{492}\text{H}_{985}$, $\text{BrMgC}_{494}\text{H}_{989}$, $\text{BrMgC}_{496}\text{H}_{993}$, $\text{BrMgC}_{498}\text{H}_{997}$, $\text{BrMgC}_{500}\text{H}_{1001}$, $\text{BrMgC}_{502}\text{H}_{1005}$, $\text{BrMgC}_{504}\text{H}_{1009}$, $\text{BrMgC}_{506}\text{H}_{1013}$, $\text{BrMgC}_{508}\text{H}_{1017}$, $\text{BrMgC}_{510}\text{H}_{1021}$, $\text{BrMgC}_{512}\text{H}_{1025}$, $\text{BrMgC}_{514}\text{H}_{1029}$, $\text{BrMgC}_{516}\text{H}_{1033}$, $\text{BrMgC}_{518}\text{H}_{1037}$, $\text{BrMgC}_{520}\text{H}_{1041}$, $\text{BrMgC}_{522}\text{H}_{1045}$, $\text{BrMgC}_{524}\text{H}_{1049}$, $\text{BrMgC}_{526}\text{H}_{1053}$, $\text{BrMgC}_{528}\text{H}_{1057}$, $\text{BrMgC}_{530}\text{H}_{1061}$, $\text{BrMgC}_{532}\text{H}_{1065}$, $\text{BrMgC}_{534}\text{H}_{1069}$, $\text{BrMgC}_{536}\text{H}_{1073}$, $\text{BrMgC}_{538}\text{H}_{1077}$, $\text{BrMgC}_{540}\text{H}_{1081}$, $\text{BrMgC}_{542}\text{H}_{1085}$, $\text{BrMgC}_{544}\text{H}_{1089}$, $\text{BrMgC}_{546}\text{H}_{1093}$, $\text{BrMgC}_{548}\text{H}_{1097}$, $\text{BrMgC}_{550}\text{H}_{1101}$, $\text{BrMgC}_{552}\text{H}_{1105}$, $\text{BrMgC}_{554}\text{H}_{1109}$, $\text{BrMgC}_{556}\text{H}_{1113}$, $\text{BrMgC}_{558}\text{H}_{1117}$, $\text{BrMgC}_{560}\text{H}_{1121}$, $\text{BrMgC}_{562}\text{H}_{1125}$, $\text{BrMgC}_{564}\text{H}_{1129}$, $\text{BrMgC}_{566}\text{H}_{1133}$, $\text{BrMgC}_{568}\text{H}_{1137}$, $\text{BrMgC}_{570}\text{H}_{1141}$, $\text{BrMgC}_{572}\text{H}_{1145}$, $\text{BrMgC}_{574}\text{H}_{1149}$, $\text{BrMgC}_{576}\text{H}_{1153}$, $\text{BrMgC}_{578}\text{H}_{1157}$, $\text{BrMgC}_{580}\text{H}_{1161}$, $\text{BrMgC}_{582}\text{H}_{1165}$, $\text{BrMgC}_{584}\text{H}_{1169}$, $\text{BrMgC}_{586}\text{H}_{1173}$, $\text{BrMgC}_{588}\text{H}_{1177}$, $\text{BrMgC}_{590}\text{H}_{1181}$, $\text{BrMgC}_{592}\text{H}_{1185}$, $\text{BrMgC}_{594}\text{H}_{1189}$, $\text{BrMgC}_{596}\text{H}_{1193}$, $\text{BrMgC}_{598}\text{H}_{1197}$, $\text{BrMgC}_{600}\text{H}_{1201}$, $\text{BrMgC}_{602}\text{H}_{1205}$, $\text{BrMgC}_{604}\text{H}_{1209}$, $\text{BrMgC}_{606}\text{H}_{1213}$, $\text{BrMgC}_{608}\text{H}_{1217}$, $\text{BrMgC}_{610}\text{H}_{1221}$, $\text{BrMgC}_{612}\text{H}_{1225}$, $\text{BrMgC}_{614}\text{H}_{1229}$, $\text{BrMgC}_{616}\text{H}_{1233}$, $\text{BrMgC}_{618}\text{H}_{1237}$, $\text{BrMgC}_{620}\text{H}_{1241}$, $\text{BrMgC}_{622}\text{H}_{1245}$, $\text{BrMgC}_{624}\text{H}_{1249}$, $\text{BrMgC}_{626}\text{H}_{1253}$, $\text{BrMgC}_{628}\text{H}_{1257}$, $\text{BrMgC}_{630}\text{H}_{1261}$, $\text{BrMgC}_{632}\text{H}_{1265}$, $\text{BrMgC}_{634}\text{H}_{1269}$, $\text{BrMgC}_{636}\text{H}_{1273}$, $\text{BrMgC}_{638}\text{H}_{1277}$, $\text{BrMgC}_{640}\text{H}_{1281}$, $\text{BrMgC}_{642}\text{H}_{1285}$, $\text{BrMgC}_{644}\text{H}_{1289}$, $\text{BrMgC}_{646}\text{H}_{1293}$, $\text{BrMgC}_{648}\text{H}_{1297}$, $\text{BrMgC}_{650}\text{H}_{1301}$, $\text{BrMgC}_{652}\text{H}_{1305}$, $\text{BrMgC}_{654}\text{H}_{1309}$, $\text{BrMgC}_{656}\text{H}_{1313}$, $\text{BrMgC}_{658}\text{H}_{1317}$, $\text{BrMgC}_{660}\text{H}_{1321}$, $\text{BrMgC}_{662}\text{H}_{1325}$, $\text{BrMgC}_{664}\text{H}_{1329}$, $\text{BrMgC}_{666}\text{H}_{1333}$, $\text{BrMgC}_{668}\text{H}_{1337}$, $\text{BrMgC}_{670}\text{H}_{1341}$, $\text{BrMgC}_{672}\text{H}_{1345}$, $\text{BrMgC}_{674}\text{H}_{1349}$, $\text{BrMgC}_{676}\text{H}_{1353}$, $\text{BrMgC}_{678}\text{H}_{1357}$, $\text{BrMgC}_{680}\text{H}_{1361}$, $\text{BrMgC}_{682}\text{H}_{1365}$, $\text{BrMgC}_{684}\text{H}_{1369}$, $\text{BrMgC}_{686}\text{H}_{1373}$, $\text{BrMgC}_{688}\text{H}_{1377}$, $\text{BrMgC}_{690}\text{H}_{1381}$, $\text{BrMgC}_{692}\text{H}_{1385}$, $\text{BrMgC}_{694}\text{H}_{1389}$, $\text{BrMgC}_{696}\text{H}_{1393}$, $\text{BrMgC}_{698}\text{H}_{1397}$, $\text{BrMgC}_{700}\text{H}_{1401}$, $\text{BrMgC}_{702}\text{H}_{1405}$, $\text{BrMgC}_{704}\text{H}_{1409}$, $\text{BrMgC}_{706}\text{H}_{1413}$, $\text{BrMgC}_{708}\text{H}_{1417}$, $\text{BrMgC}_{710}\text{H}_{1421}$, $\text{BrMgC}_{712}\text{H}_{1425}$, $\text{BrMgC}_{714}\text{H}_{1429}$, $\text{BrMgC}_{716}\text{H}_{1433}$, $\text{BrMgC}_{718}\text{H}_{1437}$, $\text{BrMgC}_{720}\text{H}_{1441}$, $\text{BrMgC}_{722}\text{H}_{1445}$, $\text{BrMgC}_{724}\text{H}_{1449}$, $\text{BrMgC}_{726}\text{H}_{1453}$, $\text{BrMgC}_{728}\text{H}_{1457}$, $\text{BrMgC}_{730}\text{H}_{1461}$, $\text{BrMgC}_{732}\text{H}_{1465}$, $\text{BrMgC}_{734}\text{H}_{1469}$, $\text{BrMgC}_{736}\text{H}_{1473}$, $\text{BrMgC}_{738}\text{H}_{1477}$, $\text{BrMgC}_{740}\text{H}_{1481}$, $\text{BrMgC}_{742}\text{H}_{1485}$, $\text{BrMgC}_{744}\text{H}_{1489}$, $\text{BrMgC}_{746}\text{H}_{1493}$, $\text{BrMgC}_{748}\text{H}_{1497}$, $\text{BrMgC}_{750}\text{H}_{1501}$, $\text{BrMgC}_{752}\text{H}_{1505}$, $\text{BrMgC}_{754}\text{H}_{1509}$, $\text{BrMgC}_{756}\text{H}_{1513}$, $\text{BrMgC}_{758}\text{H}_{1517}$, $\text{BrMgC}_{760}\text{H}_{1521}$, $\text{BrMgC}_{762}\text{H}_{1525}$, $\text{BrMgC}_{764}\text{H}_{1529}$, $\text{BrMgC}_{766}\text{H}_{1533}$, $\text{BrMgC}_{768}\text{H}_{1537}$, $\text{BrMgC}_{770}\text{H}_{1541}$, $\text{BrMgC}_{772}\text{H}_{1545}$, $\text{BrMgC}_{774}\text{H}_{1549}$, $\text{BrMgC}_{776}\text{H}_{1553}$, $\text{BrMgC}_{778}\text{H}_{1557}$, $\text{BrMgC}_{780}\text{H}_{1561}$, $\text{BrMgC}_{782}\text{H}_{1565}$, $\text{BrMgC}_{784}\text{H}_{1569}$, $\text{BrMgC}_{786}\text{H}_{1573}$, $\text{BrMgC}_{788}\text{H}_{1577}$, $\text{BrMgC}_{790}\text{H}_{1581}$, $\text{BrMgC}_{792}\text{H}_{1585}$, $\text{BrMgC}_{794}\text{H}_{1589}$, $\text{BrMgC}_{796}\text{H}_{1593}$, $\text{BrMgC}_{798}\text{H}_{1597}$, $\text{BrMgC}_{800}\text{H}_{1601}$, $\text{BrMgC}_{802}\text{H}_{1605}$, $\text{BrMgC}_{804}\text{H}_{1609}$, $\text{BrMgC}_{806}\text{H}_{1613}$, $\text{BrMgC}_{808}\text{H}_{1617}$, $\text{BrMgC}_{810}\text{H}_{1621}$, $\text{BrMgC}_{812}\text{H}_{1625}$, $\text{BrMgC}_{814}\text{H}_{1629}$, $\text{BrMgC}_{816}\text{H}_{1633}$, $\text{BrMgC}_{818}\text{H}_{1637}$, $\text{BrMgC}_{820}\text{H}_{1641}$, $\text{BrMgC}_{822}\text{H}_{1645}$, $\text{BrMgC}_{824}\text{H}_{1649}$, $\text{BrMgC}_{826}\text{H}_{1653}$, $\text{BrMgC}_{828}\text{H}_{1657}$, $\text{BrMgC}_{830}\text{H}_{1661}$, $\text{BrMgC}_{832}\text{H}_{1665}$, $\text{BrMgC}_{834}\text{H}_{1669}$, $\text{BrMgC}_{836}\text{H}_{1673}$, $\text{BrMgC}_{838}\text{H}_{1677}$, $\text{BrMgC}_{840}\text{H}_{1681}$, $\text{BrMgC}_{842}\text{H}_{1685}$, $\text{BrMgC}_{844}\text{H}_{1689}$, $\text{BrMgC}_{846}\text{H}_{1693}$, $\text{BrMgC}_{848}\text{H}_{1697}$, $\text{BrMgC}_{850}\text{H}_{1701}$, $\text{BrMgC}_{852}\text{H}_{1705}$, $\text{BrMgC}_{854}\text{H}_{1709}$, $\text{BrMgC}_{856}\text{H}_{1713}$, $\text{BrMgC}_{858}\text{H}_{1717}$, $\text{BrMgC}_{860}\text{H}_{1721}$, $\text{BrMgC}_{862}\text{H}_{1725}$, $\text{BrMgC}_{864}\text{H}_{1729}$, $\text{BrMgC}_{866}\text{H}_{1733}$, $\text{BrMgC}_{868}\text{H}_{1737}$, $\text{BrMgC}_{870}\text{H}_{1741}$, $\text{BrMgC}_{872}\text{H}_{1745}$, $\text{BrMgC}_{874}\text{H}_{1749}$, $\text{BrMgC}_{876}\text{H}_{1753}$, $\text{BrMgC}_{878}\text{H}_{1757}$, $\text{BrMgC}_{880}\text{H}_{1761}$, $\text{BrMgC}_{882}\text{H}_{1765}$, $\text{BrMgC}_{884}\text{H}_{1769}$, $\text{BrMgC}_{886}\text{H}_{1773}$, $\text{BrMgC}_{888}\text{H}_{1777}$, $\text{BrMgC}_{890}\text{H}_{1781}$, $\text{BrMgC}_{892}\text{H}_{1785}$, $\text{BrMgC}_{894}\text{H}_{1789}$, $\text{BrMgC}_{896}\text{H}_{1793}$, $\text{BrMgC}_{898}\text{H}_{1797}$, $\text{BrMgC}_{900}\text{H}_{1801}$, $\text{BrMgC}_{902}\text{H}_{1805}$, $\text{BrMgC}_{904}\text{H}_{1809}$, $\text{BrMgC}_{906}\text{H}_{1813}$, $\text{BrMgC}_{908}\text{H}_{1817}$, $\text{BrMgC}_{910}\text{H}_{1821}$, $\text{BrMgC}_{912}\text{H}_{1825}$, $\text{BrMgC}_{914}\text{H}_{1829}$, $\text{BrMgC}_{916}\text{H}_{1833}$, $\text{BrMgC}_{918}\text{H}_{1837}$, $\text{BrMgC}_{920}\text{H}_{1841}$, $\text{BrMgC}_{922}\text{H}_{1845}$, $\text{BrMgC}_{924}\text{H}_{1849}$, $\text{BrMgC}_{926}\text{H}_{1853}$, $\text{BrMgC}_{928}\text{H}_{1857}$, $\text{BrMgC}_{930}\text{H}_{1861}$, $\text{BrMgC}_{932}\text{H}_{1865}$, $\text{BrMgC}_{934}\text{H}_{1869}$, $\text{$

0.30 cm³/g. As to the percentage composition, this carrier is made up of 5 to 25% by weight of magnesium atom, 3 to 70% by weight of a halogen atom, and the balance to make up 100% by weight of organic compound and others.

As solid catalyst to be obtained by depositing on this carrier any of such transition metals as titanium, vanadium, and zirconium, which have been heretofore known as catalytic components for the polymerization of olefins manifests a highly desirable efficiency for catalysis. In the preparation of a catalyst by the use of this solid, there can be used a metal oxide such as silica or alumina or an inactive solid substance such as polyethylene, polypropylene, or some other polyolefin.

Effect of the Invention

The carrier according with the present invention consists of particles of highly satisfactory quality. The catalyst for the polymerization of an olefin prepared by using this carrier and consequently the polymer produced with this catalyst both enjoy uniformity of shape of their component particles. Thus, the catalyst manifests a high activity in the production of the polymer.

Example

The present invention will be described more specifically below with reference to examples and applied examples. Wherever the term "percent (%)" is used in these examples, it shall be construed as meaning a percent by weight unless otherwise specified.

The particle size distribution of the catalyst carrier was determined by measuring the particle size distribution of a sample with a particle sizer, a product of Malvern Corp. marketed under trademark designation of "MALVERN 3600," and carrying out a calculation of the following formula using the results of the measurement.

$$\text{PSDI} = \log \frac{\text{Particle diameter of particles accounting for 90\% of cumulative weight}}{\text{Particle diameter of particles accounting for 10\% of cumulative weight}}$$

The melt index (MI) of the polymer was determined by the method defined in ASTM D1238. The bulk density thereof was determined by the method defined in ASTM D1895-69.

The particle size distribution of the polymer was measured with the standard sieves specified by W. S. Tyler Corp. The PSDI value consequently found was reported as the index of the particle size distribution.

Example 1

Preparation of Magnesium-Containing Solid.

In a reaction vessel provided with a reflux condenser and having an inner volume of 1 liter, 12.8 g of magnesium metal chips (purity 99.5% and average particle diameter 1.6mm) and 250 ml of n-hexane were placed under an atmosphere of nitrogen gas and stirred at 68°C for one hour. Then, the magnesium metal was taken out of the resultant mixture and dried at 65°C under a vacuum to obtain preactivated magnesium metal.

Subsequently, a suspension obtained by adding to the magnesium metal 88 ml (0.53 mol) of ethyl ortho-formate $[\text{HC}(\text{OC}_2\text{H}_5)_3]$ and 0.5 ml of a methyl iodide solution of 10% of iodine was held at 55°C. The suspension and a 5 ml portion of a solution of 80 ml (0.8 mol) of n-butyl chloride in 100 ml of n-hexane added dropwise thereto were stirred for 50 minutes. Then, the remainder of the aforementioned solution was added dropwise thereto over a period of 80 minutes. The resultant combined liquid was stirred at 70°C for four hours to undergo a reaction. Consequently, there was obtained a solid reaction product.

This reaction product was washed six times each with 300 ml of n-hexane at 50°C and dried under a vacuum at 60°C for one hour, to recover 55.6 g of white powdery magnesium-containing solid. This solid was found to contain 22.5% of magnesium and 34.0% of chlorine.

Contact with 2,2,2-trichloroethanol

In a reaction vessel provided with a reflux condenser, a stirrer, and a dropping funnel and having an inner volume of 300 ml, 6.3 g of the magnesium-containing solid and 50 ml of n-heptane were placed under an atmosphere of nitrogen gas to form a suspension. This suspension was kept stirred at room temperature and a mixed solution of 2.0 ml (0.02 m.mol) of 2,2,2-trichloroethanol and 11 ml of n-heptane was added to the suspension through the dropping funnel over a period of 30 minutes. The resultant combined liquid was stirred at 80°C for one hour. The solid consequently formed therein was separated by filtration, washed six times each with 100 ml of n-hexane at room temperature, and dried under a vacuum at room temperature for one hour, to produce 8.1 g of a carrier according with the present invention. This carrier was found to contain 14.2% of

magnesium and 47.5% of chlorine and possess a specific surface area of 162 m²/g. The particle size distribution of the carrier was as shown below.

5	Diameter	Less	1.9	6.4	8.2	10.5	13.6	17.7	23.7	33.7	Above
	(μ m)	than	to	to	to	to	to	to	to	to	54.9
		1.9	6.4	8.2	10.5	13.6	17.7	23.7	33.7	54.9	
10											
	Cumulative										
	proportion	0.0	1.3	2.1	4.0	4.3	10.9	57.9	95.4	99.9	100
15	(%)										

The PSDI value of the carrier was 0.31 and the average particle diameter thereof was 23 μ m.

20 Example 2

Preparation of Magnesium-Containing Solid

By following the procedure of Example 1, 8.3 g of magnesium metal was activated. Then, a suspension obtained by adding to the magnesium metal 140 ml of n-butyl ether and 0.5 ml of methyl iodide solution of 10% of iodine was kept at 55°C. To the suspension, a solution of 38.5 ml of n-butyl chloride in 50 ml of n-butyl ether was added dropwise over a period of 50 minutes. The resultant combined liquid was stirred at 70°C for four hours to undergo a reaction. The reaction solution was kept at 55°C.

Then 55.7 ml of HC(OC₂H₅)₃ was added dropwise to the reaction solution over a period of one hour. At this time, a solid was observed to form therein. After the dropwise addition was completed, the reaction mixture was left reacting at 60°C for 15 minutes. The solid resulting from the reaction was washed six times each with 300 ml of n-hexane and dried under a vacuum at room temperature for one hour, to recover 31.6 g of a magnesium-containing solid having a magnesium content of 19.0% and a chlorine content of 28.9%.

Contact with 2,2,2-trichloroethanol

The procedure of Example 1 was repeated, except 6.3 g of the magnesium-containing solid obtained as described above was used instead. Consequently, there was obtained 8.0 g of a carrier according with the present invention. This carrier was found to contain 14.3% of magnesium and 47.7% of chlorine and possess a specific surface area of 149 m²/g. The particle size distribution of the carrier was as shown below.

40	Diameter	Less	1.9	6.4	8.2	10.5	13.6	17.7	23.7	33.7	Above
	(μ m)	than	to	to	to	to	to	to	to	to	54.9
		1.9	6.4	8.2	10.5	13.6	17.7	23.7	33.7	54.9	
45											
	Cumulative										
	Proportion	0.0	1.5	2.8	4.0	7.2	16.3	38.5	65.4	93.6	100
50	(%)										

The PSDI value of this carrier was 0.32 and the average particle diameter thereof was 22 μ m.

55 Examples 3 Through 8

Catalyst carriers according with the present invention and possessing the compositions and physical properties indicated below were obtained by following the procedure of Example 1, except the alkoxy compounds indicated below were used in the place of HC(OC₂H₅)₃.

Example	Alkoxy Compound	Composition (%)		Specific Surface Area (m ² /g)	PSDI	Average Particle Diameter (μm)	5
		Mg	Cl				
3	CH ₃ CH(OC ₂ H ₅) ₂	14.5	47.8	68	0.38	29	10
4	C(OC ₂ H ₅) ₄	14.0	55.3	94	0.45	17	
5	Si(OC ₂ H ₅) ₄	14.3	49.9	138	0.32	7	15
6	Al(OC ₂ H ₅) ₄	14.4	48.9	121	0.40	13	
7	B(OC ₂ H ₅) ₂	13.9	50.2	79	0.37	18	
8	P(OC ₂ H ₅) ₂	14.1	46.3	106	0.36	17	20

Examples 9 Through 12

Catalyst carriers according with the present invention and possessing the composition and physical properties indicated below were obtained by following the procedure of Example 1, except the halogen-containing alcohols indicated below were used in the place of 2,2,2-trichloroethanol.

Example	Halogen-Containing Alcohol	Composition (%)		Specific Surface Area (m ² /g)	PSDI	Average Particle Diameter (μm)	30
		Mg	Halogen				
9	1,1,1-Trichloro-2-propanol	19.2	42.9	141	0.39	27	35
10	2,2-Dichloro-ethanol	18.0	45.3	131	0.45	21	40
11	p-Chlorophenol	19.6	24.0	101	0.41	24	45
12	1-Bromo-2-butanol	19.4	43.1	113	0.50	26	50

Examples 13 Through 15

Catalyst carriers according with the present invention and possessing the composition and physical properties indicated below were obtained by following the procedure of Example 1, except the amount of 2,2,2-trichloroethanol (TCE) to be used was varied as shown below.

Example	Amount of TCE Used (ml)	Composition (%)		Specific Surface Area (m ² /g)	PSDI	Average Particle Diameter (μm)
		Mg	Cl			
13	1	11.7	40.1	153	0.38	23
14	4	15.2	58.1	87	0.46	29
15	6	12.1	67.7	69	0.53	31

Comparative Experiment 1

A carrier was obtained by following the procedure of Example 1, except the contact with 2,2,2-trichloroethanol was omitted. This carrier was found to have a magnesium content of 23.4%, a chlorine content of 34.0%, a specific surface area of 201 m²/g, a PSDI value of 0.38, and an average particle diameter of 17 μm.

Comparative Experiment 2

A carrier was obtained by following the procedure of Example 5, except the contact with 2,2,2-trichloroethanol was omitted. This carrier was found to have a magnesium content of 22.9%, a chlorine content of 34.0%, a specific surface area of 215 m²/g, a PSDI value of 0.32, and an average particle diameter of 6 μm.

Applied Example 1**Preparation of Catalyst Component**

In a reaction vessel provided with a reflux condenser and a stirrer and possessing an inner volume of 300 ml, 6.3 g of the carrier obtained in Example 1 and 40 ml of toluene were placed to form a suspension and the suspension and 60 ml of titanium tetrachloride added thereto were stirred at 120°C for two hours under an atmosphere of nitrogen gas. The solid substance consequently formed therein was separated by filtration and washed seven times each with 100 ml of n-hexane at room temperature, to obtain 5.7 g of a catalyst component having a titanium content of 3.8%.

Polymerization of Ethylene

In an autoclave of stainless steel (SUS 32) provided with a stirrer and having an inner volume of 1.5 liters, 10.2 mg of the catalyst component obtained as described above, 0.7 m.mol of triisobutyl aluminum, and 390 g of isobutane were placed under an atmosphere of nitrogen gas to produce a system for polymerization and the polymerization system was heated to 85°C. Then, hydrogen was introduced therein until the partial pressure of hydrogen reached 20 kg/cm² and ethylene was introduced therein until the partial pressure of ethylene reached 5.0 kg/cm². Ethylene was continuously fed to the autoclave, there to be polymerized, for 60 minutes at such a rate as to maintain the total pressure in the system constant. After the polymerization was completed, the contents of the autoclave were purged of the solvent and the unaltered ethylene. The white powdery polymer consequently formed was removed from the autoclave and dried under a vacuum at 70°C for 10 hours, to obtain 245 g of powdery polyethylene having an MI value of 3.3 g/10 minutes and a bulk density of 0.38 g/cm³ [Kc (catalytic activity) 24,000 g of polymer per g of catalyst component]. The average particle diameter of this polyethylene was 750 μm and the PSDI value thereof was 0.33.

Applied Examples 2 Through 17

Preparation of catalyst components and polymerization of ethylene were carried out by following the procedure of Applied Example 1, except the carriers obtained in Examples 2 through 15, Comparative Experiment 1, and Comparative Experiment 2 were used in the place of the carrier obtained in Example 1. The results were as shown below.

Applied Example	Carrier	Kc (g/g-cat.)	MI (g/10 min.)	Bulk Density (g/cm ³)	PSDI	
2	Example 2	26600	4.3	0.39	0.30	5
3	Example 3	22800	2.8	0.37	0.33	10
4	Example 4	17300	3.3	0.37	0.40	
5	Example 5	11900	3.8	0.39	0.29	
6	Example 6	14700	3.5	0.40	0.38	15
7	Example 7	17500	4.1	0.37	0.33	
8	Example 8	16000	2.9	0.38	0.31	
9	Example 9	19000	3.1	0.35	0.35	20
10	Example 10	20200	3.3	0.38	0.45	
11	Example 11	18400	3.7	0.38	0.41	
12	Example 12	22500	2.9	0.38	0.49	25
13	Example 13	18700	3.3	0.39	0.35	
14	Example 14	26200	3.5	0.40	0.33	30
15	Example 15	25300	3.8	0.40	0.33	
16	Comparative	19900	2.1	0.20	0.69	
	Experiment 1					35
17	Experiment 2	11600	2.1	0.39	0.31	

Brief Description of the Drawing

Figure 1 is a flow chart illustrating the method according with the present invention.

Claims

1. An olefin polymerization catalyst carrier obtained by contacting

I. a magnesium-containing solid obtained by contact of (A) magnesium metal, (B) a halogenated hydrocarbon represented by the general formula RX , wherein R stands for an alkyl, aryl or cycloalkyl group having 1 to 20 carbon atoms and X for a halogen atom, and (C) a compound of the general formula $X \frac{1}{n} M(OR^1)_{m-n}$, wherein X^1 stands for a hydrogen atom, a halogen atom, or a hydrocarbon group having 1 to 20 carbon atoms, M stands for a boron, carbon, silicon, aluminum, or phosphorus atom, R^1 represents a hydrocarbon group having 1 to 20 carbon atoms, m for the valency of said atom M, and $m > n \geq 0$, with

II. a halogen-containing alcohol.

2. The catalyst carrier in accordance with claim 1 wherein the Mg metal and the halogenated hydrocarbon are contacted so as to provide a Grignard reagent.

3. The catalyst carrier in accordance with claim 1 or 2 wherein R is an alkyl group having from 1 to 8 carbon atoms, cyclohexyl or an aryl group having 6 or 7 carbon atoms and X is chlorine, the alkoxy compound is selected from the group consisting of $HC(OC_2H_5)_3$, $CH_3CH(OC_2H_5)_2$, $C(OC_2H_5)_4$, $Si(OC_2H_5)_4$, $Al(OC_2H_5)_4$, $B(OC_2H_5)_3$ and $P(OC_2H_5)_3$, and the halogenated alcohol is selected from 2,2,2-trichloroethanol, 1,1,1-trichloro 2-propanol, 2,2-dichloroethanol, p-chlorophenol and 1-bromo-2-butanol.

4. The catalyst carrier in accordance with claim 3 wherein the halogenated hydrocarbon is n-butyl

chloride and the compound represented by the formula $X \frac{1}{n} M(OR^1)_{m-n}$ is one of orthoformate or ethyl orthoformate.

5 5. The catalyst carrier in accordance with any of claims 1-4 wherein the alkoxy compound is contacted with the magnesium compound in the amount such that the number of OR^1 groups in the alkoxy compound per magnesium atom exceeds 1.

6. The catalyst carrier in accordance with claim 5 wherein the ratio of alkoxy groups to the magnesium is in the range of 3 to 5.

7. The catalyst carrier in accordance with any of claims 1-6 wherein the amount of halogenated alcohol contacted with the magnesium is in the range of 0.05 to 20 per gram atom of magnesium in the solid.

10 8. A method for the production of an olefin polymerization catalyst carrier comprising contacting

15 I. a magnesium-containing solid obtained by contact of (A) magnesium metal, (B) a halogenated hydrocarbon represented by the general formula RX , wherein R represents an alkyl, aryl or cycloalkyl group having 1 to 20 carbon atoms and X for a halogen atom, and (C) a compound of the general formula $X \frac{1}{n} M(OR^1)_{m-n}$, wherein X^1 represents a hydrogen atom, a halogen atom, or a hydrocarbon group having 1 to 20 carbon atoms, M stands for a boron, carbon, silicon, aluminum, or phosphorus atom, R^1 represents a hydrocarbon group having 1 to 20 carbon atoms, m for the valency of said atom M , and $m > n \geq 0$, with

II. a halogen-containing alcohol.

20

25

30

35

40

45

50

55

60

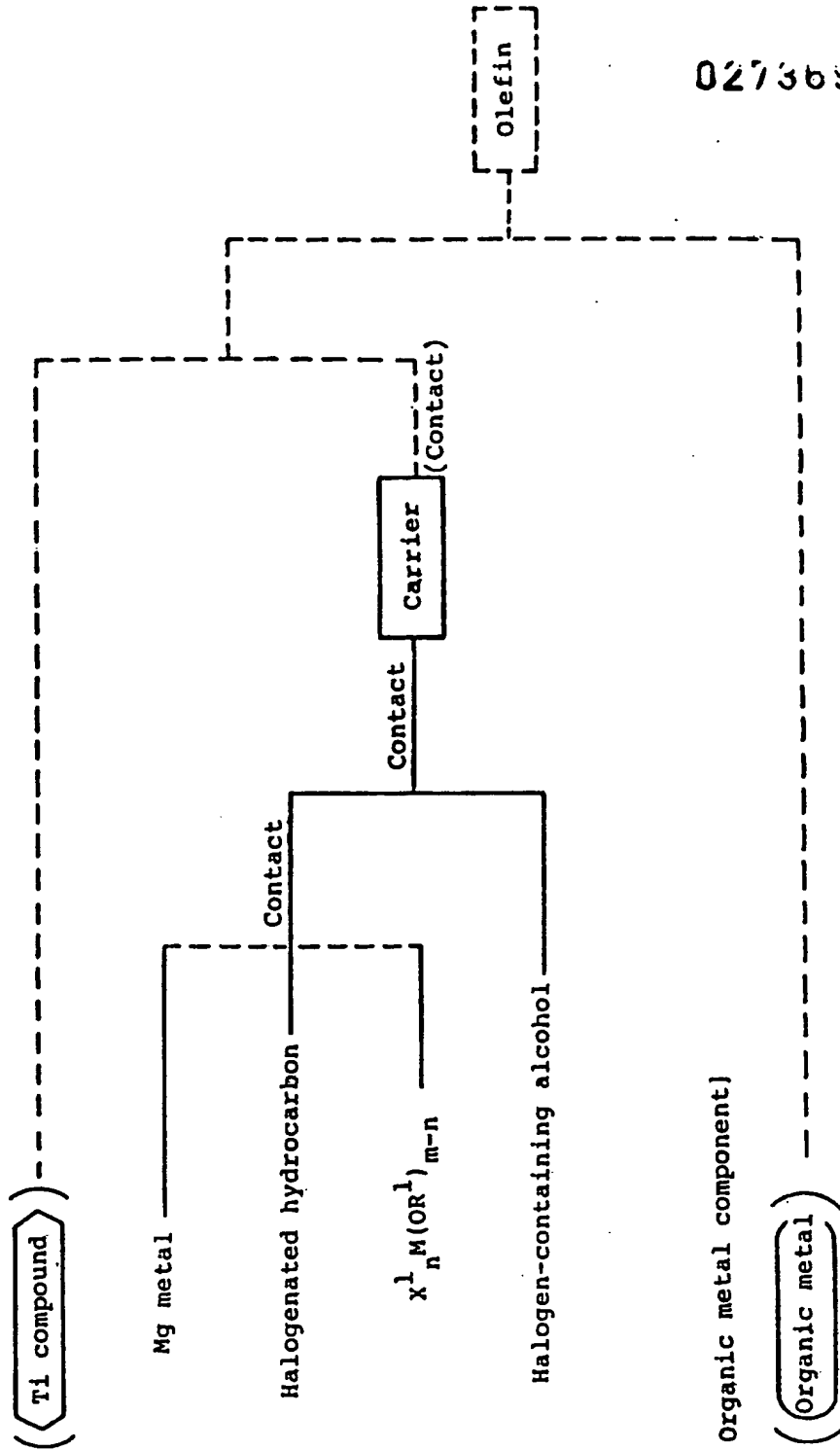
65

07.04.88

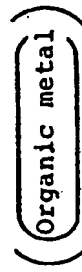
0273695

FIG.1

[(A). Transition metal component]



[(B) Organic metal component]





European Patent
Office

EUROPEAN SEARCH REPORT

Application Number

EP 87 31 1333

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
Y	EP-A-0 053 899 (TOA NENRYO) * Claims 1-9 *	1	C 08 F 4/02 C 08 F 4/64 C 08 F 10/00
Y	EP-A-0 173 472 (TOA NENRYO) * Claims 1-10 *	1	
			TECHNICAL FIELDS SEARCHED (Int. Cl.4)
			C 08 F
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 25-02-1988	Examiner DE ROECK R.G.
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	